

Chapter11 Review Gases Answer Key

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questions in the space provided. 1. c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 . L. (b) divided by the mass of 1 mol. (d) divided by 22.4 . L. 2. c For the expression $V = nRT$,

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REVIEW Gases

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Modern Chemistry 97 Gases CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. ____ The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2. ____ For the expression , P

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The volume of a fixed mass of gas is DIRECTLY proportional to the Kelvin Temperature, when pressure is held constant
Equation: $V_1/T_1 = V_2/T_2$ $P = \text{Constant}$. Combined Gas Law.
expresses the relationship between pressure, volume and temperature of a fixed amount of gas
Equation: $P_1V_1/T_1 = P_2V_2/T_2$. The Ideal Gas Law.

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